

BASF Coatings AG
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Clearcoat material and its use to produce clearcoats
and multicoat color and/or effect coating systems

The present invention relates to a novel clearcoat
5 material, in particular an aqueous clearcoat material.
The present invention further relates to a process for
preparing the novel clearcoat material. The present
invention additionally relates to the use of the novel
clearcoat material to produce clearcoats and multicoat
10 color and/or effect finishes, especially for motor
vehicles. The present invention relates, furthermore,
to novel clearcoats and multicoat systems and to the
primed or unprimed substrates coated therewith,
especially motor vehicle bodies.

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Clearcoat materials for producing clearcoats are
customary and known. They comprise one-component (1K),
two component (2K) or multicomponent (3K, 4K)
clearcoats, powder clearcoats, powder slurry clear-
20 coats, and UV-curable clearcoats.

One-component (1K), two-component (2K) or
multicomponent (3K, 4K) clearcoats are known from
patents DE-A-42 04 518, US-A-5,474,811, US-A-5,356,669,
25 US-A-5,605,965, WO 94/10211, WO 94/10212, WO 94/10213,
EP-A-0 594 068, EP-A-0 594 071, EP-A-0 594 142,

EP-A-0 604 992, WO 94/22969, EP-A-0 596 460, and
WO 92/22615.

One-component (1K) clearcoat materials comprise, as is
5 known, hydroxyl-containing binders and crosslinking
agents such as blocked polyisocyanates, tris(alkoxy-
carbonylamino)triazines and/or amino resins. In another
variant, they comprise as binders polymers containing
10 pendant carbamate and/or allophanate groups, and
carbamate- and/or allophanate-modified amino resins as
crosslinking agents (cf. US-A-5,474,811,
US-A-5,356,669, US-A-5,605,965, WO 94/10211,
WO 94/10212, WO 94/10213, EP-A-0 594 068,
EP-A-0 594 071, and EP-A-0 594 142).

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Two-component (2K) or multicomponent (3K, 4K)
clearcoats comprise as essential constituents, as is
known, hydroxyl-containing binders and polyisocyanates
as crosslinking agents, which are stored separately
20 prior to their use.

Powder clearcoats are known, for example, from German
Patent DE-A-42 22 194 or from the product information
sheet from BASF Lacke + Farben AG, "Pulverlacke"
25 [Powder coatings], 1990.

Powder clearcoats comprise as essential constituents, as is known, binders containing epoxide groups, and polycarboxylic acids as crosslinking agents.

5 Powder slurry clearcoats are known, for example, from U.S. Patent US-A-4,268,542 and German Patent Applications DE-A-195 18 392.4 and DE-A-196 13 547 and are described in German Patent Application DE-A-198 14 471.7, unpublished at the priority date of
10 the present specification.

Powder slurry clearcoats comprise, as is known, powder clearcoats in dispersion in an aqueous medium.

15 UV-curable clearcoats are disclosed, for example, in patents EP-A-0 540 884, EP-A-0 568 967, and US-A-4,675,234.

20 Predominantly, these clearcoat materials comprise acrylate copolymers as their binders.

The known clearcoat materials may be used in accordance with the wet-on-wet technique, which is employed with preference in automotive OEM finishing; i.e., basecoat
25 materials are applied to a primed or unprimed substrate to give a basecoat film which, however, is not cured but instead is only initially dried and overcoated with

the clearcoat film, after which the two films are jointly cured.

Clearcoat materials, especially aqueous clearcoat
5 materials, have numerous advantages which render them attractive for industrial utilization.

The acrylate copolymers themselves may be prepared by well-known polymerization techniques in bulk, solution
10 or emulsion. Polymerization techniques for preparing acrylate copolymers, especially polyacrylate resins, are general knowledge and are widely described (cf., e.g., Houben-Weyl, Methoden der organischen Chemie, 4th Edition, Volume 14/1, pages 24 to 255. (1961)).

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Further examples of suitable copolymerization techniques for preparing acrylate copolymers are described in patents DE-A-197 09 465, DE-C-197 09 476, DE-A-28 48 906, DE-A-195 24 182, EP-A-0 554 783, 20 EP-B-0 650 979, WO 95/27742, DE-A-38 41 540, and WO 82/02387.

Suitable reactors for the copolymerization processes are the customary and known stirred vessels, cascades
25 of stirred vessels, tube reactors, loop reactors, and Taylor reactors, as described for example in patents DE-B-1 071 241 and EP-A-0 498 583 or in the article by

K. Kataoka in Chemical Engineering Science, Volume 50,
No. 9, 1995, 1409 to 1416.

The free-radical polymerization used to prepare the
5 acrylate copolymers, however, is often very exothermic
and difficult to control. The implications of this fact
for the reaction regime are that it is necessary to
avoid high monomer concentrations and/or the batch
mode, where the entirety of the monomers is introduced
10 as an initial charge in an aqueous medium, emulsified
and subsequently polymerized to completion. Even the
tailoring of defined molecular weights, molecular
weight distributions, and other properties frequently
causes difficulties. The tailoring of a certain profile
15 of properties in the acrylate copolymers, however, is
of great importance for their use as binders in
clearcoat materials, especially aqueous clearcoat
materials, since by this means it is possible to
influence the profile of performance properties of the
20 clearcoat materials in a direct way.

There has therefore been no lack of attempts to control
the free-radical copolymerization of olefinically
unsaturated monomers.

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For instance, International Patent Application
WO 98/01478 describes a process in which the
copolymerization is conducted in the presence of a

free-radical initiator and of a thiocarbonylthio compound as chain transfer agent.

International Patent Application WO 92/13903 describes 5 a process for preparing copolymers having a low molecular weight by means of free-radical chain polymerization in the presence of a group transfer agent containing a carbon-sulfur double bond. These compounds act not only as chain transfer agents but 10 also as growth regulators, so that only low molecular weight copolymers result.

International Patent Application WO 96/15157 discloses 15 a process for preparing copolymers having a comparatively narrow molecular weight distribution, in which a monomer is reacted with a vinyl-terminated macromonomer in the presence of a free-radical initiator.

20 Furthermore, International Patent Application WO 98/37104 discloses the preparation of acrylate copolymers having defined molecular weights by means of free-radical polymerization in the presence of a chain transfer agent containing a C-C double bond and 25 containing radicals which activate this double bond in terms of the free-radical addition reaction of monomers.

Despite the significant progress in this area, there is still a lack of a universally applicable process of controlled free-radical polymerization which in a 5 simple manner provides chemically structured polymers, especially acrylate copolymers, and by means of which it is possible to tailor the profile of properties of the polymers in respect of their use in clearcoat materials, especially aqueous clearcoat materials, 10 which are used to produce clearcoats and multicoat color and/or effect coating systems.

It is an object of the present invention to provide new clearcoat materials, especially aqueous clearcoat 15 materials, which are outstandingly suitable for producing clearcoats, especially as part of multicoat color and/or effect coating systems, and as an advantageous alternative to the conventional clearcoat materials. A further object of the present invention is 20 to propose new processes for preparing the clearcoat materials, permitting their profile of properties to be varied in a simple way and to be adapted precisely to the profiles of properties of the other coats of the multicoat systems. The aim is to realize these objects 25 in a simple manner by tailoring the profile of properties of the clearcoat materials, in particular through the use of chemically structured polymers obtainable by means of controlled free-radical

polymerization. The new clearcoats and new multicoat
color and/or effect coating systems which result with
the aid of these new clearcoat materials should possess
at least the good profile of properties of the known
5 clearcoats and multicoat systems, and in fact should
preferably exceed them. In particular, they are
intended to possess outstanding optical quality,
intercoat adhesion and condensation resistance and to
exhibit no cracking of the basecoats (mud cracking),
10 flow defects or surface structures.

Accordingly, we have found the novel use of a copolymer
(A) in a clearcoat material, especially an aqueous
clearcoat material, which is used to produce clearcoats
15 and multicoat color and/or effect coating systems,
where the copolymer (A) is preparable by free-radical
polymerization of

- a) at least one olefinically unsaturated monomer and
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- b) at least one olefinically unsaturated monomer
different than the olefinically unsaturated
monomer (a) and of the general formula I

in which the radicals R^1 , R^2 , R^3 and R^4 each independently of one another are hydrogen atoms or substituted or unsubstituted alkyl, cycloalkyl, alkylcycloalkyl, cycloalkylalkyl, aryl, alkylaryl, cycloalkylaryl, arylalkyl or arylcycloalkyl radicals, with the proviso that at least two of the variables R^1 , R^2 , R^3 and R^4 are substituted or unsubstituted aryl, arylalkyl or arylcycloalkyl radicals, especially substituted or unsubstituted aryl radicals;

in an aqueous medium.

In the text below, the novel use of the copolymer (A) is referred to as "use in accordance with the invention".

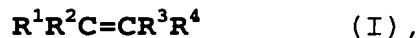
We have also found the novel clearcoat material, especially aqueous clearcoat material, comprising

20 (A) as binder, or one of the binders, at least one copolymer preparable by free-radical polymerization of

25 a) at least one olefinically unsaturated monomer and

b) at least one olefinically unsaturated monomer different than the olefinically unsaturated monomer (a) and of the general formula I

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in which the radicals R^1 , R^2 , R^3 and R^4 each independently of one another are hydrogen atoms or substituted or unsubstituted alkyl, cycloalkyl, alkylcycloalkyl, cycloalkylalkyl, aryl, alkylaryl, cycloalkylaryl, arylalkyl or arylcycloalkyl radicals, with the proviso that at least two of the variables R^1 , R^2 , R^3 and R^4 are substituted or unsubstituted aryl, arylalkyl or arylcycloalkyl radicals, especially substituted or unsubstituted aryl radicals,

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in an aqueous medium;

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and

(B) at least one crosslinking agent.

25 In the text below, the novel clearcoat material is referred to as the "clearcoat material of the invention".

We have also found the novel process for producing a multicoat color and/or effect coating system ML on a primed or unprimed substrate by

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(I) preparing a basecoat film by applying a basecoat material to the substrate,

(II) drying the basecoat film,

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(III) preparing a clearcoat film by applying a clearcoat material to the basecoat film, and

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(IV) jointly curing the basecoat film and the clearcoat film to give the basecoat BL and the clearcoat KL,

or

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(I) preparing a surfacer film by applying a surfacer to the substrate,

(II) curing the surfacer film to give the surfacer coat FL,

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(III) preparing a basecoat film by applying a basecoat material to the surfacer coat FL,

(IV) drying the basecoat film,

5 (V) preparing a clearcoat film by applying a clearcoat material to the basecoat film, and

10 (VI) jointly curing the basecoat film and the clearcoat film to give the basecoat BL and the clearcoat KL,

15 in which the clearcoat material of the invention is used as clearcoat material.

20 In the text below, the novel process for producing a multicoat color and/or effect coating system ML on a primed or unprimed substrate is referred to as the "process of the invention".

25 In the light of the prior art it was surprising that the object on which the present invention was based could be achieved by means of the use in accordance with the invention, the clearcoat material of the invention, especially the aqueous clearcoat material of the invention, and the process of the invention. A particular surprise was that the clearcoats KL and multicoat systems ML of the invention have outstanding properties even at comparatively low baking temperatures. In particular, they are of outstanding

optical quality, possess good intercoat adhesion and condensation resistance, and exhibit no cracking in the basecoat (mud cracking), flow defects or surface structures.

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In accordance with the invention, at least one copolymer (A) is used as the binder (A), or one of the binders (A), in the clearcoat material of the invention.

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In accordance with the invention, the copolymer (A) is prepared by free-radical polymerization of at least one olefinically unsaturated monomer (a) and at least one olefinically unsaturated monomer (b) which is different 15 than the monomer (a).

Examples of suitable monomers (a) are

a1) essentially acid-group-free (meth)acrylic esters 20 such as (meth)acrylic alkyl or cycloalkyl esters having up to 20 carbon atoms in the alkyl radical, especially methyl, ethyl, propyl, n-butyl, sec-butyl, tert-butyl, hexyl, ethylhexyl, stearyl and lauryl acrylate or methacrylate; cycloaliphatic (meth)acrylic esters, especially cyclohexyl, 25 isobornyl, dicyclopentadienyl, octahydro-4,7-methano-1H-indenemethanol acrylate or tert-butylcyclohexyl (meth)acrylate; (meth)acrylic

oxaalkyl esters or oxacycloalkyl esters such as ethyltriglycol (meth)acrylate and methoxy-oligoglycol (meth)acrylate having a molecular weight Mn of preferably 550, or other ethoxylated and/or propoxylated hydroxyl-free (meth)acrylic acid derivatives. These may contain minor amounts of (meth)acrylic alkyl or cycloalkyl esters of higher functionality, such as the di(meth)acrylates of ethylene glycol, propylene glycol, diethylene glycol, dipropylene glycol, butylene glycol, 1,5-pentanediol, 1,6-hexanediol, octahydro-4,7-methano-1H-indenedimethanol or 1,2-, 1,3- or 1,4-cyclohexanediol; trimethylolpropane di- or tri(meth)acrylate; or pentaerythritol di-, tri- or tetra(meth)acrylate. For the purposes of the present invention, minor amounts of monomers of higher functionality in this case are to be understood as amounts which do not lead to crosslinking or gelling of the copolymers (A).

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a2) Monomers which carry per molecule at least one hydroxyl group, amino group, alkoxyethylamino group or imino group and are essentially free from acid groups, such as hydroxyalkyl esters of acrylic acid, methacrylic acid or another alpha,beta-olefinically unsaturated carboxylic acid, which derive from an alkylene glycol esterified with the acid, or which are obtainable

by reacting the alpha,beta-olefinically unsaturated carboxylic acid with an alkylene oxide, especially hydroxyalkyl esters of acrylic acid, methacrylic acid, ethacrylic acid, crotonic acid, maleic acid, fumaric acid or itaconic acid in which the hydroxyalkyl group contains up to 20 carbon atoms, such as 2-hydroxyethyl, 2-hydroxypropyl, 3-hydroxypropyl, 3-hydroxybutyl, 4-hydroxybutyl acrylate, methacrylate, ethacrylate, crotonate, maleate, fumarate or itaconate; or hydroxycycloalkyl esters such as 1,4-bis(hydroxymethyl)cyclohexane, octahydro-4,7-methano-1H-indenedimethanol or methylpropanediol monoacrylate, monomethacrylate, monoethacrylate, monocrotonate, monomaleate, monofumarate or monoitaconate; or reaction products of cyclic esters, such as epsilon-caprolactone, for example, and these hydroxyalkyl or hydroxycycloalkyl esters; or olefinically unsaturated alcohols such as allyl alcohol or polyols such as trimethylolpropane monoallyl or diallyl ether or pentaerythritol monoallyl, diallyl or triallyl ether (as far as these monomers (a2) of higher functionality are concerned, the comments made above relating to the monomers (a1) of higher functionality apply analogously); N,N-dimethylaminoethyl acrylate, N,N-diethylaminoethyl methacrylate, allylamine or N-methyliminoethyl

acrylate or N,N-di(methoxymethyl)aminoethyl acrylate and methacrylate or N,N-di(butoxymethyl)-aminopropyl acrylate and methacrylate; monomers of this kind are used preferably to prepare 5 selfcrosslinking constituents (A).

a3) Monomers which carry per molecule at least one acid group which can be converted to the corresponding acid anion group, such as acrylic acid, methacrylic acid, ethacrylic acid, crotonic acid, maleic acid, fumaric acid or itaconic acid; olefinically unsaturated sulfonic or phosphonic acids or their partial esters; or mono(meth)-acryloyloxyethyl maleate, succinate or phthalate. 10

15 a4) Vinyl esters of alpha-branched monocarboxylic acids having 5 to 18 carbon atoms in the molecule. The branched monocarboxylic acids can be obtained by reacting formic acid or carbon monoxide and water with olefins in the presence of a liquid, 20 strongly acidic catalyst; the olefins may be cracking products of paraffinic hydrocarbons, such as mineral oil fractions, and may comprise both branched and straight-chain acyclic and/or 25 cycloaliphatic olefins. The reaction of such olefins with formic acid or, respectively, with carbon monoxide and water produces a mixture of carboxylic acids in which the carboxyl groups are

located predominantly on a quaternary carbon atom. Examples of other olefinic starting materials are propylene trimer, propylene tetramer and diisobutylene. Alternatively, the vinyl esters 5 (a4) may be prepared in a conventional manner from the acids, by reacting, for example, the acid with acetylene. Particular preference, owing to their ready availability, is given to using vinyl esters of saturated aliphatic monocarboxylic acids having 10 9 to 11 carbon atoms that are branched on the alpha carbon atom, especially Versatic® acids.

a5) Reaction products of acrylic acid and/or methacrylic acid with the glycidyl ester of an 15 alpha-branched monocarboxylic acid having 5 to 18 carbon atoms per molecule, especially a Versatic® acid, or, instead of the reaction product, an equivalent amount of acrylic acid and/or methacrylic acid which is then reacted during or 20 after the polymerization reaction with the glycidyl ester of an alpha-branched monocarboxylic acid having 5 to 18 carbon atoms per molecule, especially a Versatic® acid.

25 a6) Cyclic and/or acyclic olefins such as ethylene, propylene, 1-butene, 1-pentene, 1-hexene, cyclohexene, cyclopentene, norbornene, butadiene,

isoprene, cyclopentadiene and/or dicyclo-
pentadiene.

a7) (Meth)acrylamides such as (meth)acrylamide,
5 N-methyl-, N,N-dimethyl-, N-ethyl-, N,N-diethyl-,
N-propyl-, N,N-dipropyl-, N-butyl-, N,N-dibutyl-,
N-cyclohexyl-, N,N-cyclohexylmethyl- and/or
N-methylol-, N,N-dimethylol-, N-methoxymethyl-,
10 N,N-di(methoxymethyl)-, N-ethoxymethyl- and/or
N,N-di(ethoxyethyl)-(meth)acrylamide. Monomers of
the last-mentioned kind are used in particular to
prepare selfcrosslinking binders (A).

a8) Monomers containing epoxide groups, such as the
15 glycidyl ester of acrylic acid, methacrylic acid,
ethacrylic acid, crotonic acid, maleic acid,
fumaric acid and/or itaconic acid.

a9) Vinylaromatic hydrocarbons such as styrene, alpha-
20 alkylstyrenes, especially alpha-methylstyrene,
and/or vinyltoluene; vinylbenzoic acid (all
isomers), N,N-diethylaminostyrene (all isomers),
alpha-methylvinylbenzoic acid (all isomers), N,N-
diethylamino-alpha-methylstyrene (all isomers)
25 and/or p-vinylbenzenesulfonic acid.

a10) Nitriles such as acrylonitrile and/or
methacrylonitrile.

11) Vinyl compounds, especially vinyl halides and/or vinylidene dihalides such as vinyl chloride, vinyl fluoride, vinylidene dichloride or vinylidene difluoride; N-vinylamides such as vinyl-N-methylformamide, N-vinylcaprolactam, 1-vinylimidazole or N-vinylpyrrolidone; vinyl ethers such as ethyl vinyl ether, n-propyl vinyl ether, isopropyl vinyl ether, n-butyl vinyl ether, isobutyl vinyl ether and/or vinyl cyclohexyl ether; and/or vinyl esters such as vinyl acetate, vinyl propionate, vinyl butyrate, vinyl pivalate and/or the vinyl ester of 2-methyl-2-ethyl-heptanoic acid.

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12) Allyl compounds, especially allyl ethers and allyl esters such as allyl methyl, ethyl, propyl or butyl ether or allyl acetate, propionate or butyrate.

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13) Polysiloxane macromonomers having a number-average molecular weight M_n of from 1000 to 40,000 and having on average from 0.5 to 2.5 ethylenically unsaturated double bonds per molecule; especially polysiloxane macromonomers having a number-average molecular weight M_n of from 2000 to 20,000, with particular preference from 2500 to 10,000 and, in particular, from 3000 to 7000 and having on

average from 0.5 to 2.5, preferably from 0.5 to 1.5, ethylenically unsaturated double bonds per molecule, as are described in DE 38 07 571 A1 on pages 5 to 7, in DE 37 06 095 A1 in columns 3 to 5, in EP 0 358 153 B1 on pages 3 to 6, in US 4,754,014 A in columns 5 to 9, in DE 44 21 823 A1 or in International Patent Application WO 92/22615 A on page 12 line 18 to page 18 line 10.

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a14) Acryloxsilane-containing vinyl monomers, preparable by reacting hydroxyl-functional silanes with epichlorohydrin and then reacting the reaction product with (meth)acrylic acid and/or 15 with hydroxyalkyl and/or hydroxycycloalkyl esters of (meth)acrylic acid (cf. monomers a2).

Each of the abovementioned monomers (a1) to (a14) may be polymerized on their own with the monomers (b). In 20 accordance with the invention, however, it is advantageous to use at least two monomers (a), since by this means it is possible to vary the profile of properties of the resulting copolymers (A) very widely, in a particularly advantageous manner, and to tailor 25 said profile of properties to the particular intended use of the clearcoat material. In particular, it is possible in this way to incorporate into the copolymers (A) functional groups by means of which the copolymers

(A) become hydrophilic, so that they may be dissolved or dispersed in aqueous media. It is also possible to incorporate functional groups (afg) capable of entering into thermal crosslinking reactions with the 5 complementary functional groups (cfg), described below, of any crosslinking agents (C) used. It is also possible to attach functional groups which give the copolymer (A) selfcrosslinking properties, such as N-methylol groups or N-alkoxymethyl groups.

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In accordance with the invention, very particular advantages result if the monomers (a) used comprise the monomers (a1) and (a2) and also, if desired, (a3).

15 In accordance with the invention, the monomers (b) used comprise compounds of the general formula I.

In the general formula I, the radicals R^1 , R^2 , R^3 and R^4 each independently of one another are hydrogen atoms or 20 substituted or unsubstituted alkyl, cycloalkyl, alkylcycloalkyl, cycloalkylalkyl, aryl, alkylaryl, cycloalkylaryl, arylalkyl or arylcycloalkyl radicals, with the proviso that at least two of the variables R^1 , R^2 , R^3 and R^4 are substituted or unsubstituted aryl, 25 arylalkyl or arylcycloalkyl radicals, especially substituted or unsubstituted aryl radicals.

Examples of suitable alkyl radicals are methyl, ethyl, propyl, isopropyl, n-butyl, isobutyl, tert-butyl, amyl, hexyl or 2-ethylhexyl.

5 Examples of suitable cycloalkyl radicals are cyclobutyl, cyclopentyl or cyclohexyl.

Examples of suitable alkylcycloalkyl radicals are methylenecyclohexane, ethylenecyclohexane or propane-10 1,3-diylcyclohexane.

Examples of suitable cycloalkylalkyl radicals are 2-, 3- or 4-methyl-, -ethyl-, -propyl- or -butylcyclohex-1-yl.

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Examples of suitable aryl radicals are phenyl, naphthyl or biphenylyl, preferably phenyl and naphthyl, and especially phenyl.

20 Examples of suitable alkylaryl radicals are benzyl or ethylene- or propane-1,3-diylbenzene.

Examples of suitable cycloalkylaryl radicals are 2-, 3- or 4-phenylcyclohex-1-yl.

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Examples of suitable arylalkyl radicals are 2-, 3- or 4-methyl-, -ethyl-, -propyl- or -butylphen-1-yl.

Examples of suitable arylcycloalkyl radicals are 2-, 3- or 4-cyclohexylphen-1-yl.

- 5 The above-described radicals R^1 , R^2 , R^3 and R^4 may be substituted. The substituents used may comprise electron-withdrawing or electron-donating atoms or organic radicals.
- 10 Examples of suitable substituents are halogen atoms, especially chlorine and fluorine, nitrile groups, nitro groups, partially or fully halogenated, especially chlorinated and/or fluorinated, alkyl, cycloalkyl, alkylcycloalkyl, cycloalkylalkyl, aryl, alkylaryl, 15 cycloalkylaryl, arylalkyl and arylcycloalkyl radicals, including those exemplified above, especially tert-butyl; aryloxy, alkyloxy and cycloalkyloxy radicals, especially phenoxy, naphthoxy, methoxy, ethoxy, propoxy, butyloxy or cyclohexyloxy; arylthio, alkylthio 20 and cycloalkylthio radicals, especially phenylthio, naphthylthio, methylthio, ethylthio, propylthio, butylthio or cyclohexylthio; hydroxyl groups; and/or primary, secondary and/or tertiary amino groups, especially amino, N-methylamino, N-ethylamino, 25 N-propylamino, N-phenylamino, N-cyclohexylamino, N,N-dimethylamino, N,N-diethylamino, N,N-dipropylamino, N,N-diphenylamino, N,N-dicyclohexylamino, N-cyclohexyl-N-methylamino and N-ethyl-N-methylamino.

Examples of monomers (b) whose use is particularly preferred in accordance with the invention are diphenylethylene, dinaphthaleneethylene, cis- or 5 trans-stilbene, vinylidenebis(4-N,N-dimethylamino- benzene), vinylidenebis(4-aminobenzene), and vinylidene-bis(4-nitrobenzene).

In accordance with the invention, the monomers (b) may 10 be used individually or as a mixture of at least two monomers (b).

In terms of the reaction regime and the properties of the resultant copolymers (A), especially the acrylate 15 copolymers (A), diphenylethylene is of very particular advantage and is therefore used with very particular preference in accordance with the invention.

The monomers (a) and (b) to be used in accordance with 20 the invention are reacted with one another in the presence of at least one free-radical initiator to form the copolymer (A). Examples of initiators which can be used are: dialkyl peroxides, such as di-tert-butyl peroxide or dicumyl peroxide; hydroperoxides, such as 25 cumene hydroperoxide or tert-butyl hydroperoxide; peresters, such as tert-butyl perbenzoate, tert-butyl perpivalate, tert-butyl per-3,5,5-trimethylhexanoate or tert-butyl per-2-ethylhexanoate; potassium, sodium or

ammonium peroxodisulfate; azodinitriles such as azobisisobutyronitrile; C-C-cleaving initiators such as benzpinacol silyl ethers; or a combination of a nonoxidizing initiator with hydrogen peroxide.

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It is preferred to add comparatively large amounts of free-radical initiator, the proportion of the initiator in the reaction mixture being, based in each case on the overall amount of the monomers (a) and of the 10 initiator, with particular preference from 0.5 to 50% by weight, with very particular preference from 1 to 20% by weight, and in particular from 2 to 15% by weight.

15 Preferably, the weight ratio of initiator to the monomers (b) is from 4:1 to 1:4, with particular preference from 3:1 to 1:3, and in particular from 2:1 to 1:2. Further advantages result if the initiator is used in excess within the stated limits.

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The free-radical copolymerization is preferably conducted in the apparatus mentioned at the outset, especially stirred vessels or Taylor reactors, the Taylor reactors being designed such that the conditions 25 of Taylor flow are met over the entire reactor length, even if the kinematic viscosity of the reaction medium alters greatly, and in particular increases, owing to the copolymerization.

In accordance with the invention, the copolymerization is conducted in an aqueous medium.

5 The aqueous medium comprises essentially water. The aqueous medium may include minor amounts of the below-detailed crosslinking agents (C), reactive diluents (G), coatings additives (H) and/or organic solvents (I) and/or other dissolved solid, liquid or gaseous organic
10 and/or inorganic substances of low and/or high molecular mass, especially surface-active substances, provided these do not adversely affect, or even inhibit, the copolymerization. In the context of the present invention, a "minor amount" is to be understood
15 as an amount which does not remove the aqueous character of the aqueous medium.

Alternatively, the aqueous medium may comprise straight water.

20 The copolymerization is preferably conducted in the presence of at least one base. Particular preference is given to low molecular mass bases such as sodium hydroxide solution, potassium hydroxide solution,
25 ammonia, diethanolamine, triethanolamine, mono-, di- and triethylamine, and/or dimethylethanolamine, especially ammonia and/or di- and/or triethanolamine.

The copolymerization is advantageously conducted at temperatures above room temperature and below the lowest decomposition temperature of the monomers used

5 in each case, preference being given to a chosen temperature range of from 10 to 150°C, with very particular preference from 70 to 120°C, and in particular from 80 to 110°C.

10 When using particularly volatile monomers (a) and/or (b), the copolymerization may also be conducted under pressure, preferably under from 1.5 to 3000 bar, with particular preference from 5 to 1500 bar, and in particular from 10 to 1000 bar.

15 In terms of the molecular weight distributions, there are no restrictions whatsoever imposed on the constituent (A). Advantageously, however, the copolymerization is conducted so as to give a molecular weight distribution M_w/M_n , measured by gel permeation chromatography using polystyrene as standard, of ≤ 4 , with particular preference ≤ 2 , and in particular ≤ 1.5 , and in certain cases even ≤ 1.3 . The molecular weights of the constituents (A) may be controlled

20 within wide limits by the choice of ratio of monomer (a) to monomer (b) to free-radical initiator. In this context, the amount of monomer (b) in particular

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determines the molecular weight, specifically such that, the higher the proportion of monomer (b), the lower the resultant molecular weight.

- 5 The constituent (A) resulting from the copolymerization is obtained as a mixture with the aqueous medium, generally in the form of a dispersion. In this form it can be processed further directly or else used as a macroinitiator for further reaction with at least one
- 10 further monomer (a) in a second stage (ii). The constituent (A) resulting in the first stage (i), however, may also be isolated as a solid and then reacted further.
- 15 The further reaction in accordance with the stage (ii) is preferably conducted under the standard conditions for a free-radical polymerization, it being possible for suitable solvents (H) and/or reactive diluents (F) to be present. Stages (i) and (ii) in the context of
- 20 the process of the invention may be conducted separately from one another, both spatially and temporally. In addition, however, stages (i) and (ii) may also be carried out in succession in one reactor. For this purpose, the monomer (b) is first reacted with
- 25 at least one monomer (a), completely or partially depending on the desired application and the desired properties, after which at least one further monomer (a) is added and the mixture is subjected to free-

radical polymerization. In another embodiment, at least two monomers (a) are used from the start, the monomer (b) being first reacted with one of the at least two monomers (a) and then the resultant reaction product 5 (A) being reacted, above a certain molecular weight, with the further monomer (a) as well.

Depending on reaction regime, it is possible in accordance with the invention to prepare endgroup-10 functionalized polymers, block, multiblock and gradient copolymers, star polymers, graft copolymers, and branched copolymers as constituents (A).

The copolymer (A) may include at least one, preferably 15 at least two, functional groups (afg) which are able to enter into thermal crosslinking reactions with complementary functional groups (bfg) of the crosslinking agents (B) described below. The functional groups (afg) may be introduced into the constituent (A) 20 by way of the monomers (a) or may be introduced following its synthesis, by means of polymer-analogous reactions.

Examples of suitable complementary reactive functional 25 groups (afg) and (bfg) which enter into crosslinking reactions, for use in accordance with the invention, are summarized in the following overview. In the overview, the variable R^5 is substituted or

unsubstituted alkyl, cycloalkyl, alkylcycloalkyl, cycloalkylalkyl, aryl, alkylaryl, cycloalkylaryl, arylalkyl or arylcycloalkyl radicals; the variables R⁶ and R⁷ are identical or different alkyl, cycloalkyl, 5 alkylcycloalkyl or cycloalkylalkyl radicals, or are linked with one another to form an aliphatic or heteroaliphatic ring. Examples of suitable radicals of this kind are those listed above in connection with the radicals R¹, R², R³ and R⁴.

Overview: Examples of complementary functional groups (afg) and (bfg) in the

constituent (A) and crosslinking agent (B)

5

or

crosslinking agent (B) and constituent (A)

-SH

-C(O)-OH

-NH₂

-C(O)-O-C(O)-

-OH

-NCO

-O-(CO)-NH-(CO)-NH₂

-NH-C(O)-OR⁵

-O-(CO)-NH₂

-CH₂-OH

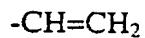
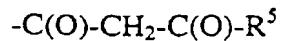
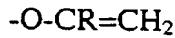
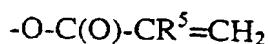
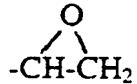
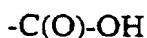
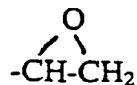
-CH₂-O-CH₃

-NH-C(O)-CH(-C(O)OR⁵)₂

-NH-C(O)-CH(-C(O)OR⁵)(-C(O)-R⁵)

-NH-C(O)-NR⁶R⁷

= Si(OR⁵)₂



The selection of the respective complementary groups (afg) and (bfg) is guided on the one hand by the consideration that, during storage, they should not enter into any unwanted reactions and/or should not disrupt or inhibit curing, if appropriate, with actinic radiation, and on the other hand by the temperature range within which thermal curing is to take place.

10

In this context, especially with regard to heat-sensitive substrates such as plastics, it is of advantage in accordance with the invention to choose a temperature range which does not exceed 100°C, and in particular does not exceed 80°C. In the light of these

boundary conditions, complementary functional groups which have proven advantageous are hydroxyl groups and isocyanate groups, or carboxyl groups and epoxy groups, which are therefore employed with preference, in accordance with the invention, in the clearcoat materials of the invention that are present as two-component or multi-component systems. Particular advantages result if the hydroxyl groups are used as functional groups (afg) and the isocyanate groups as functional groups (bfg).

If higher crosslinking temperatures, for example from 100°C to 160°C, may be employed, which is preferred in accordance with the invention, suitable clearcoat materials also include one-component systems, in which the functional groups (afg) are preferably thio, amino, hydroxyl, carbamate, allophanate, carboxyl and/or (meth)acrylate groups, but especially hydroxyl groups, and the functional groups (bfg) are preferably anhydride, carboxyl, epoxy, blocked isocyanate, urethane, methylol, methylol ether, siloxane, amino, hydroxyl and/or beta-hydroxyalkylamide groups.

The proportion of the copolymer (A) for use in accordance with the invention in the novel clearcoat material may vary very widely. Advantageously, the proportion is from 1 to 90, preferably from 2 to 80, with particular preference from 3 to 75, and in

particular from 4 to 70% by weight, based on the overall solids content of the clearcoat material of the invention.

5 The clearcoat material of the invention may further comprise at least one customary and known binder (A) with at least one functional group (afg). Examples of suitable customary and known binders (A) are linear and/or branched and/or block, comb and/or random
10 poly(meth)acrylates or acrylate copolymers, polyesters, alkyds, amino resins, polyurethanes, acrylated polyurethanes, acrylated polyesters, polylactones, polycarbonates, polyethers, epoxy resin-amine adducts, (meth)acrylate diols, partially hydrolyzed polyvinyl
15 esters or polyureas, which contain said functional groups (afg). If used, their proportion in the clearcoat material of the invention is preferably from 1 to 50, preferably from 2 to 40, with particular preference from 3 to 30, and in particular from four to
20 25% by weight, based in each case on the overall solids content of the clearcoat material of the invention.

The clearcoat material further comprises at least one crosslinking agent (B) which contains at least two, 25 especially three, of the complementary functional groups (bfg) described in detail above.

Where the clearcoat material comprises a two-component or multicomponent system, polyisocyanates and/or polyepoxides, but especially polyisocyanates, are used as crosslinking agents (B).

5

Examples of suitable polyisocyanates (B) are organic polyisocyanates, especially so-called paint polyisocyanates, having free isocyanate groups attached to aliphatic, cycloaliphatic, araliphatic and/or aromatic 10 moieties. Preference is given to polyisocyanates having from 2 to 5 isocyanate groups per molecule and viscosities of from 100 to 10,000, preferably from 100 to 5000, and in particular from 100 to 2000 mPa.s (at 23°C). If desired, small amounts of organic solvent 15 (H), preferably from 1 to 25% by weight based on straight polyisocyanate, may be added to the polyisocyanates in order to make it easier to incorporate the isocyanate and, if appropriate, to reduce the viscosity of the polyisocyanate to a level 20 within the abovementioned ranges. Examples of suitable solvent additives to the polyisocyanates are ethoxyethyl propionate, amyl methyl ketone, and butyl acetate. Furthermore, the polyisocyanates (B) may have been hydrophilically or hydrophobically modified in a 25 customary and known manner.

Examples of suitable polyisocyanates (B) are described, for example, in "Methoden der organischen Chemie",

Houben-Weyl, Volume 14/2, 4th Edition, Georg Thieme Verlag, Stuttgart, 1963, pages 61 to 70, and by W. Siefken, Liebigs Annalen der Chemie, Volume 562, pages 75 to 136.

5

Further examples of suitable polyisocyanates (B) are polyisocyanates containing isocyanurate, biuret, allophanate, iminoxadiazinedione, urethane, urea and/or uretdione groups. Polyisocyanates containing 10 urethane groups, for example, are prepared by reacting some of the isocyanate groups with polyols, such as trimethylolpropane and glycerol, for example. Preference is given to the use of aliphatic or cycloaliphatic polyisocyanates, especially hexa-15 methylene diisocyanate, dimerized and trimerized hexamethylene diisocyanate, isophorone diisocyanate, 2-isocyanatopropylcyclohexyl isocyanate, dicyclohexylmethane 2,4'-diisocyanate, dicyclohexylmethane 4,4'-diisocyanate, or 1,3-bis(isocyanatomethyl)cyclohexane 20 (BIC), diisocyanates derived from dimeric fatty acids, as marketed under the commercial designation DDI 1410 by Henkel, 1,8-diisocyanato-4-isocyanatomethyloctane, 1,7-diisocyanato-4-isocyanatomethylheptane or 1-isocyanato-2-(3-isocyanatopropyl)cyclohexane, or mixtures 25 of these polyisocyanates.

Examples of suitable polyepoxides (B) are all known aliphatic and/or cycloaliphatic and/or aromatic

polyepoxides, examples being those based on bisphenol A or bisphenol F. Other suitable examples of polyepoxides are the polyepoxides available commercially under the designations Epikote® from Shell, Denacol® from Nagase Chemicals Ltd., Japan, such as Denacol EX-411 (pentaerythritol polyglycidyl ether), Denacol EX-321 (trimethylolpropane polyglycidyl ether), Denacol EX-512 (polyglycerol polyglycidyl ether), and Denacol EX-521 (polyglycerol polyglycidyl ether).

10

In the case of the one-component systems, the crosslinking agents (B) used react at relatively high temperatures with the functional groups (afg) of the binders (A) to build up a three-dimensional network. Of course, such crosslinking agents (B) may be used as well in the multicomponent systems, in minor amounts. In the context of the present invention, a "minor amount" is a fraction which does not disrupt, let alone prevent, the principal crosslinking reaction.

15

Examples of suitable crosslinking agents (B) of this kind are blocked polyisocyanates. Examples of suitable polyisocyanates for preparing the blocked polyisocyanates are those described above.

20

Examples of suitable blocking agents are the blocking agents known from U.S. Patent US-A-4,444,954, such as

- i) phenols such as phenol, cresol, xyleneol, nitrophenol, chlorophenol, ethylphenol, t-butylphenol, hydroxybenzoic acid, esters of this acid, 5 or 2,5-di-tert-butyl-4-hydroxytoluene;
- ii) lactams, such as ϵ -caprolactam, δ -valerolactam, γ -butyrolactam or β -propiolactam;
- 10 iii) active methylenic compounds, such as diethyl malonate, dimethyl malonate, ethyl or methyl acetoacetate, or acetylacetone;
- iv) alcohols such as methanol, ethanol, n-propanol, 15 isopropanol, n-butanol, isobutanol, t-butanol, n-amyl alcohol, t-amyl alcohol, lauryl alcohol, ethylene glycol monomethyl ether, ethylene glycol monoethyl ether, ethylene glycol monobutyl ether, diethylene glycol monomethyl ether, diethylene 20 glycol monoethyl ether, propylene glycol monomethyl ether, methoxymethanol, glycolic acid, glycolic esters, lactic acid, lactic esters, methylolurea, methylolmelamine, diacetone alcohol, ethylenechlorohydrin, ethylenebromohydrin, 1,3- 25 dichloro-2-propanol, 1,4-cyclohexyldimethanol or acetocyanohydrin;

v) mercaptans such as butyl mercaptan, hexyl mercaptan, t-butyl mercaptan, t-dodecyl mercaptan, 2-mercaptobenzothiazole, thiophenol, methylthiophenol or ethylthiophenol;

5

vi) acid amides such as acetoanilide, acetoanisidine, acrylamide, methacrylamide, acetamide, stearamide or benzamide;

10 vii) imides such as succinimide, phthalimide or maleimide;

15 viii) amines such as diphenylamine, phenylnaphthylamine, xylidine, N-phenylxylidine, carbazole, aniline, naphthylamine, butylamine, dibutylamine or butyl-phenylamine;

ix) imidazoles such as imidazole or 2-ethylimidazole;

20 x) ureas such as urea, thiourea, ethyleneurea, ethylenethiourea or 1,3-diphenylurea;

xi) carbamates such as phenyl N-phenylcarbamate or 2-oxazolidone;

25

xii) imines such as ethyleneimine;

xiii) oximes such as acetone oxime, formaldoxime, acetaldoxime, acetoxime, methyl ethyl ketoxime, diisobutyl ketoxime, diacetyl monoxime, benzophenone oxime or chlorohexanone oximes;

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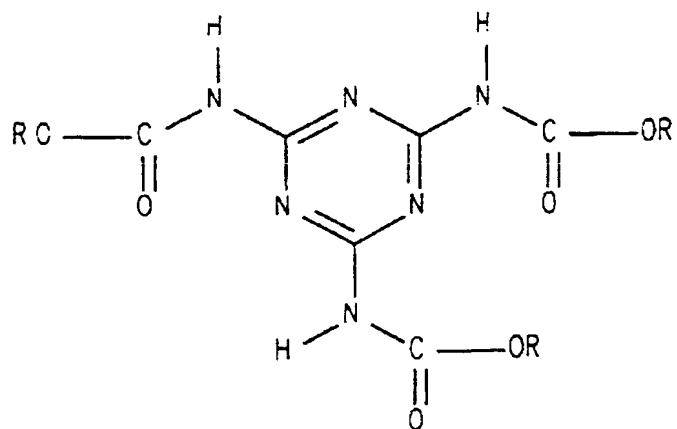
xiv) salts of sulfurous acid such as sodium bisulfite or potassium bisulfite;

10 xv) hydroxamic esters such as benzyl methacrylohydroxamate (BMH) or allyl methacrylohydroxamate; or

15 xvi) substituted pyrazoles, ketoximes, imidazoles or triazoles; and also

mixtures of these blocking agents, especially dimethylpyrazole and triazoles, malonic esters and acetoacetic esters or dimethylpyrazole and succinimide.

20 As crosslinking agent (B) it is also possible to use tris(alkoxycarbonylamino)triazines (TACT) of the general formula



Examples of suitable tris(alkoxycarbonylamino)triazines

(B) are described in patents US-A-4,939,213,

5 US-A-5,084,541, and EP-A-0 624 577. Use is made in particular of the tris(methoxy-, tris(butoxy- and/or tris(2-ethylhexoxycarbonylamino)triazines.

The methyl butyl mixed esters, the butyl 2-ethylhexyl
10 mixed esters, and the butyl esters are of advantage. They have the advantage over the straight methyl ester of better solubility in polymer melts, and also have less of a tendency to crystallize out.

15 Further examples of suitable crosslinking agents (B) are amino resins, examples being melamine resins, guanamine resins, and urea resins. For further details, reference is made to Römp Lexikon Lacke und Druckfarben, Georg Thieme Verlag, 1998, page 29, "Amino
20 resins", and the textbook "Lackadditive" [Coatings additives] by Johan Bieleman, Wiley-VCH, Weinheim, New

York, 1998, pages 242 ff., or the book "Paints, Coatings and Solvents", second, completely revised edition, D. Stoye and W. Freitag (eds.), Wiley-VCH, Weinheim, New York, 1998, pages 80 ff. Also suitable
5 are the customary and known amino resins some of whose methyolol and/or methoxymethyl groups have been defunctionalized by means of carbamate or allophanate groups. Crosslinking agents of this kind are described in patents US-A-4,710,542 and EP-B-0 245 700 and also
10 in the article by B. Singh and coworkers, "Carbamylmethylated Melamines, Novel Crosslinkers for the Coatings Industry" in Advanced Organic Coatings Science and Technology Series, 1991, Volume 13, pages 193 to 207.

15

Further examples of suitable crosslinking agents (B) are beta-hydroxyalkylamides such as N,N,N',N'-tetrakis(2-hydroxyethyl)adipamide or N,N,N',N'-tetra-kis(2-hydroxypropyl)adipamide.

20

Further examples of suitable crosslinking agents (B) are siloxanes, especially siloxanes containing at least one trialkoxy- or dialkoxysilane group.

25 Further examples of suitable crosslinking agents (B) are polyanhydrides, especially polysuccinic anhydride.

Further examples of suitable crosslinking agents (B) are compounds containing on average at least two groups amenable to transesterification, examples being reaction products of malonic diesters and polyisocyanates or 5 reaction products of monoisocyanates with esters and partial esters of malonic acid with polyhydric alcohols, as described in European Patent EP-A-0 596 460.

10 The amount of the crosslinking agents (B) in the clearcoat material may vary widely and is guided in particular, firstly, by the functionality of the crosslinking agents (B) and, secondly, by the number of crosslinking functional groups (afg) which are present 15 in the binder (A), and also by the target crosslinking density. The skilled worker is therefore able to determine the amount of the crosslinking agents (B) on the basis of his or her general knowledge in the art, possibly with the aid of simple rangefinding 20 experiments. Advantageously, the crosslinking agent (B) is present in the clearcoat material of the invention in an amount of from 1 to 60, preferably from 2 to 50, with particular preference from 3 to 45, and in particular from 4 to 40% by weight, based in each case 25 on the overall solids content of the clearcoat material of the invention. It is further advisable here, in the case of its use, to choose the amounts of crosslinking agent (B) and binder (A) such that in the clearcoat

material the ratio of functional groups (bfg) in the crosslinking agent (B) to the functional groups (afg) in the binder (A) is from 2:1 to 1:2, preferably from 1.5:1 to 1:1.5, with particular preference from 1.2:1 to 1:1.2, and in particular from 1.1:1 to 1:1.1.

If the clearcoat material of the invention is to be curable not only thermally but also with actinic radiation, especially UV radiation and/or electron beams (dual cure), it comprises at least one constituent (C) which is curable with actinic radiation. If, however, the clearcoat material of the invention is to be curable predominantly (dual cure) or exclusively with actinic radiation, it must comprise a constituent (C).

Suitable constituents (C) are in principle all oligomeric and polymeric compounds that are curable with actinic radiation, especially UV radiation and/or electron beams, said compounds being commonly used in the field of UV curable or electron beam curable clearcoat materials.

Radiation curable binders are used advantageously as constituents (C). Examples of suitable radiation curable binders (C) are (meth)acrylic-functional (meth)acrylic copolymers, polyether acrylates, polyester acrylates, unsaturated polyesters, epoxy acrylates, urethane acrylates, amino acrylates,

melamine acrylates, silicone acrylates, isocyanato acrylates, and the corresponding methacrylates. It is preferred to use binders (C) that are free from aromatic structural units. Preference is therefore 5 given to the use of urethane (meth)acrylates and/or polyester (meth)acrylates, with particular preference aliphatic urethane acrylates.

Where constituents (C) are used, they are present in 10 the clearcoat material in an amount of preferably from 1 to 60, more preferably from 1.5 to 50, with particular preference from 2 to 40, and in particular from 2.5 to 30% by weight, based in each case on the overall solids content of the clearcoat material of the 15 invention.

The clearcoat material of the invention may further comprise at least one photoinitiator (D). If the clearcoat material or the coats produced from it are to 20 be crosslinked additionally with UV radiation in the context of the process of the invention, it is generally necessary to use a photoinitiator (D). Where used, it is present in the clearcoat material of the invention in fractions of preferably from 0.01 to 10, 25 more preferably from 0.1 to 8, and in particular from 0.5 to 6% by weight, based in each case on the overall solids content of the clearcoat material of the invention.

Examples of suitable photoinitiators (D) are those of the Norrish II type, whose mechanism of action is based on an intramolecular variant of the hydrogen abstraction reactions as occur diversely in photochemical reactions (reference may be made here, by way of example, to Römpf Chemie Lexikon, 9th, expanded and revised edition, Georg Thieme Verlag, Stuttgart, Vol. 4, 1991) or cationic photoinitiators (reference 5 may be made here, by way of example, to Römpf Lexikon Lacke und Druckfarben, Georg Thieme Verlag, Stuttgart, 1998, pages 444 to 446), especially benzophenones, benzoins or benzoin ethers, or phosphine oxides. It is also possible to use, for example, the products 10 available commercially under the names Irgacure® 184, Irgacure® 1800 and Irgacure® 500 from Ciba Geigy, Grenocure® MBF from Rahn, and Lucirin® TPO from BASF 15 AG.

20 In addition to the photoinitiators (D), use may be made of customary sensitizers (D) such as anthracene in effective amounts.

25 Furthermore, the clearcoat material may comprise at least one thermal crosslinking initiator (E). At from 80 to 120°C, these initiators form free radicals which start the crosslinking reaction. Examples of thermally

labile free-radical initiators are organic peroxides, organic azo compounds or C-C-cleaving initiators such as dialkyl peroxides, peroxocarboxylic acids, peroxodisulfides, peroxide esters, hydroperoxides, 5 ketone peroxides, azo dinitriles, or benzpinacol silyl ethers. Particular preference is given to C-C-cleaving initiators, since their thermal cleavage does not produce any gaseous decomposition products which might lead to defects in the coating film. Where used, their 10 amounts are generally from 0.01 to 10, preferably from 0.05 to 8, and in particular from 0.1 to 5% by weight, based in each case on the clearcoat material of the invention.

15 Moreover, the clearcoat material may comprise at least one reactive diluent (F) curable thermally and/or with actinic radiation.

Examples of suitable thermally crosslinkable reactive 20 diluents (F) are branched, cyclic and/or acyclic C₉-C₁₆ alkanes functionalized with at least two hydroxyl groups; preferably dialkyloctanediois, especially the positionally isomeric diethyloctanediois.

25 Further examples of suitable thermally crosslinkable reactive diluents (F) are oligomeric polyols obtainable from oligomeric intermediates themselves obtained by metathesis reactions from acyclic monoolefins and

cyclic monoolefins, by hydroformylation and subsequent hydrogenation; examples of suitable cyclic monoolefins are cyclobutene, cyclopentene, cyclohexene, cyclooctene, cycloheptene, norbornene or 7-oxanorbornene; 5 examples of suitable acyclic monoolefins are present in hydrocarbon mixtures obtained in petroleum processing by cracking (C₅ cut); examples of suitable oligomeric polyols for use in accordance with the invention have a hydroxyl number (OHN) of from 200 to 450, a number- 10 average molecular weight Mn of from 400 to 1000, and a mass-average molecular weight Mw of from 600 to 1100.

Further examples of suitable thermally crosslinkable reactive diluents (F) are hyperbranched compounds 15 containing a tetrafunctional central group, derived from ditrimethylolpropane, diglycerol, ditrimethylol-ethane, pentaerythritol, tetrakis(2-hydroxyethyl) methane, tetrakis(3-hydroxypropyl)methane or 2,2-bishydroxymethyl-1,4-butanediol (homopentaerythritol). 20 These reactive diluents may be prepared in accordance with the customary and known methods of preparing hyperbranched and dendimeric compounds. Suitable synthesis methods are described, for example, in patents WO 93/17060 or WO 96/12754 or in the book by 25 G.R. Newkome, C.N. Moorefield and F. Vögtle, "Dendritic Molecules, Concepts, Syntheses, Perspectives", VCH, Weinheim, New York, 1996.

Further examples of suitable reactive diluents (F) are polycarbonate diols, polyester polyols, poly(meth)acrylate diols, and hydroxyl-containing polyadducts.

5

Examples of suitable reactive solvents which may be used as reactive diluents (F) are butyl glycol, 2-methoxypropanol, n-butanol, methoxybutanol, n-propanol, ethylene glycol monomethyl ether, ethylene glycol monoethyl ether, ethylene glycol monobutyl ether, diethylene glycol monomethyl ether, diethylene glycol monoethyl ether, diethylene glycol diethyl ether, diethylene glycol monobutyl ether, trimethylolpropane, ethyl 2-hydroxypropionate and 3-methyl-3-methoxybutanol, and also derivatives based 15 on propylene glycol, e.g., isopropoxypropanol.

Examples of the active diluents (F) used that may be crosslinked with actinic radiation are polysiloxane 20 macromonomers, (meth)acrylic acid and other esters thereof, maleic acid and its esters, including monoesters, vinyl acetate, vinyl ethers, vinylureas, and the like. Examples that may be mentioned include alkylene glycol di(meth)acrylate, polyethylene glycol di(meth)acrylate, 1,3-butanediol di(meth)acrylate, vinyl (meth)acrylate, allyl (meth)acrylate, glycerol tri(meth)acrylate, trimethylolpropane tri(meth)acrylate, trimethylolpropane di(meth)acrylate, tripropylene 25

glycol diacrylate, styrene, vinyltoluene,
divinylbenzene, pentaerythritol tri(meth)acrylate,
pentaerythritol tetra(meth)acrylate, dipropylene glycol
di(meth)acrylate, hexanediol di(meth)acrylate,
5 ethoxyethoxyethyl acrylate, N-vinylpyrrolidone,
phenoxyethyl acrylate, dimethylaminoethyl acrylate,
hydroxyethyl (meth)acrylate, butoxyethyl acrylate,
isobornyl (meth)acrylate, dimethylacrylamide, and
dicyclopentyl acrylate, and also the long-chain linear
10 diacrylates described in EP-A-0 250 631 and having a
molecular weight of from 400 to 4000, preferably from
600 to 2500. The acrylate groups may also, for example,
be separated by a polyoxybutylene structure. Further
candidates for use are 1,12-dodecyl diacrylate and the
15 reaction product of 2 mol of acrylic acid with one mole
of a dimeric fatty alcohol having generally 36 carbon
atoms. Mixtures of the abovementioned monomers are also
suitable.

20 Preferred for use as reactive diluents (F) are mono-
and/or diacrylates, such as isobornyl acrylate,
hexanediol diacrylate, tripropylene glycol diacrylate,
Laromer® 8887 from BASF AG, and Actilane® 423 from
Akcros Chemicals Ltd., GB. Particular preference is
25 given to the use of isobornyl acrylate, hexanediol
diacrylate, and tripropylene glycol diacrylate.

Where used, the reactive diluents (F) are employed in an amount of preferably from 1 to 70, with particular preference from 2 to 65, and in particular from 3 to 50% by weight, based in each case on the overall solids 5 content of the clearcoat material of the invention.

The clearcoat material of the invention may comprise customary coatings additives (G) in effective amounts. The critical factor here is that the additives (G) do 10 not adversely affect or even destroy the clarity and transparency of the clearcoat material of the invention. Advantageously, the additives (G) are not volatile under the processing application conditions of the clearcoat material of the invention.

15

Examples of suitable additives (G) are

- UV absorbers;
- free-radical scavengers;
- 20 - crosslinking catalysts such as dibutyltin dilaurate or lithium decanoate;
- slip additives;
- polymerization inhibitors;

- defoamers;
- dryers;
- antiskinning agents;
- neutralizing agents such as ammonia or 5 dimethylethanolamine;
- emulsifiers, especially nonionic emulsifiers such as alkoxylated alkanols and polyols, phenols and alkylphenols or anionic emulsifiers such as alkali metal salts or ammonium salts of alkanecarboxylic acids, alkanesulfonic acids, and sulfo acids of 10 alkoxylated alkanols and polyols, phenols and alkylphenols;
- wetting agents such as siloxanes, fluorine compounds, carboxylic monoesters, phosphoric esters, polyacrylic acids and their copolymers, or polyurethanes; 15
- adhesion promoters such as tricyclodecanedimethanol;
- leveling agents;
- film-forming auxiliaries such as cellulose derivatives;

- transparent fillers based on silica, alumina or zirconium oxide; for further details reference is made to Römpf Lexikon "Lacke und Druckfarben", Georg Thieme Verlag, Stuttgart, 1998, pages 250 to 252;
- 5 - sag control agents such as ureas, modified ureas and/or silicas, as described for example in the references EP-A-0 192 304, DE-A-23 59 923, DE-A-18 05 693, WO 94/22968, DE-C-27 51 761, WO 97/12945 or "farbe + lack", 11/1992, pages 829 ff.;
- 10 - rheology control additives such as those known from patents WO 94/22968, EP-A-0 276 501, EP-A-0 249 201 or WO 97/12945; crosslinked polymeric microparticles, such as disclosed, for example, in EP-A-0 008 127; 15 inorganic phyllosilicates such as aluminum-magnesium silicates, sodium-magnesium and sodium-magnesium-fluorine-lithium phyllosilicates of the montmorillonite type; silicas such as Aerosils; or synthetic polymers containing ionic and/or associative groups such as polyvinyl alcohol, poly(meth)acrylamide, poly(meth)acrylic acid, polyvinylpyrrolidone, styrene-maleic anhydride or ethylene-maleic anhydride copolymers and their derivatives, or hydrophobically modified ethoxylated 20 urethanes or polyacrylates;
- 25

- flame retardants; and/or

- biocides.

Further examples of suitable coatings additives (H) are described in the textbook "Lackadditive" by Johan 5 Bieleman, Wiley-VCH, Weinheim, New York, 1998.

The clearcoat material preferably comprises these additives (G) in amounts of up to 40, with particular preference up to 30, and in particular up to 20% by 10 weight, based in each case on the overall solids content of the coating material.

Not least, the clearcoat materials of the invention, especially in the case of nonaqueous clearcoat 15 materials, may comprise from 1 to 70, preferably from 2 to 60, and in particular from 3 to 50% by weight, based on the application-ready clearcoat material of the invention, of water-miscible and water-immiscible organic solvents (H), such as aliphatic, aromatic 20 and/or cycloaliphatic hydrocarbons such as toluene or methylcyclohexane or decalin; alkyl esters of acetic acid or propionic acid; alkanols such as ethanol; ketones such as methyl isobutyl ketone; glycol ethers; glycol ether esters and/or ethers such as 25 tetrahydrofuran. In the context of the present

invention it is also possible to use carbon dioxide as solvent (H).

The clearcoat material of the invention may be present 5 in different forms.

Thus, given an appropriate choice of its above-described constituents (A) and (B), and, if appropriate, of at least one of its constituents (A; 10 customary and known binders), (C), (D), (E), (F) and/or (G), it may be present in the form of a liquid clearcoat material which is essentially free from organic solvents and/or water (100% system).

15 However, the clearcoat material may also comprise a solution or dispersion of the above-described constituents in organic solvents (H) and/or water. It is a further advantage of the clearcoat material of the invention that in this case it is possible to establish 20 solids contents of up to more than 80% by weight, based on the clearcoat material.

Moreover, given an appropriate choice of its above-described constituents, the clearcoat material of the 25 invention may be a powder coating material. For this purpose, the constituent (B) may have been microencapsulated if it is a polyisocyanate. This powder coating material may then, if desired, be

dispersed in water, to give a powder slurry clearcoat material.

The clearcoat material of the invention may be a two-
5 component or multicomponent system in which at least constituent (B) is stored separately from the other constituents and is not added to them until shortly before use. In this case, the clearcoat material of the invention may also be aqueous, the constituent (B) 10 preferably being present in a component comprising a solvent (H). This variant of the clearcoat material of the invention is employed in particular for automotive refinishing.

15 Furthermore, the clearcoat material of the invention may be part of a so-called mixer system or modular system, as described, for example, in patent DE-A-41 10 520, EP-A-0 608 773, EP-A-0 614 951, or EP-A-0 471 972.

20

Preferably, the clearcoat material of the invention is in the form of an aqueous solution, a dispersion and/or an emulsion, in particular a dispersion, since in this case there is no need to isolate the copolymer (A) for 25 use in accordance with the invention.

The preparation of the clearcoat material of the invention from its constituents (A) and (B) and also,

if appropriate, at least one of its constituents (A; customary and known binders), (C), (D), (E), (F), (G) and/or (H) has no special features but instead takes place in a customary and known manner by mixing the 5 constituents in appropriate mixing equipment such as stirred vessels, dissolvers or extruders in accordance with the techniques suitable for the preparation of the respective clearcoat materials.

10 The clearcoat material of the invention is used to produce the clearcoat KL and multicoat system ML of the invention on primed or unprimed substrates.

Suitable coating substrates are fundamentally all 15 surfaces which are undamaged by curing of the coatings present thereon using heat and, if appropriate, actinic radiation; examples are metals, plastics, wood, ceramic, stone, textile, fiber composites, leather, glass, glass fibers, glass wool, rock wool, mineral-20 and resin-bound building materials, such as plasterboard and cement slabs or roof tiles, and also assemblies of these materials.

Accordingly, the clearcoat material of the invention is 25 fundamentally also suitable for applications outside of automotive finishing, for example, in industrial coating, including coil coating and container coating. In the context of industrial coatings it is suitable

for coating virtually all parts and articles for private or industrial use, such as radiators, domestic appliances, small metal parts, wheel hubs or rims. Furthermore, the clearcoat material of the invention is 5 also suitable for varnishing furniture.

In the case of electrically conductive substrates it is possible to use primers which are produced in a customary and known manner from electrodeposition 10 coating materials. For this purpose, both anodic and cathodic electrodeposition coating materials are suitable, but especially cathodics.

Using the clearcoat material of the invention it is 15 also possible in particular to coat primed or unprimed plastics such as, for example, ABS, AMMA, ASA, CA, CAB, EP, UF, CF, MF, MPF, PF, PAN, PA, PE, HDPE, LDPE, LLDPE, UHMWPE, PET, PMMA, PP, PS, SB, PUR, PVC, RF, SAN, PBT, PPE, POM, PUR-RIM, SMC, BMC, PP-EPDM, and UP 20 (abbreviated codes in accordance with DIN 7728P1). The plastics to be coated may of course also be polymer blends, modified plastics, or fiber reinforced plastics. It is also possible to employ the plastics commonly used in vehicle construction, especially motor 25 vehicle construction. Unfunctionalized and/or nonpolar substrate surfaces may be subjected prior to coating in a known manner to a pretreatment, such as with a plasma

or by flaming, or may be provided with a water-based primer.

The application of the clearcoat material of the invention has no special features in terms of its method but instead may take place by any customary application method, such as spraying, knife coating, brushing, flow coating, dipping, or rolling. It is preferred to employ spray application methods, such as compressed air spraying, airless spraying, high-speed rotation, electrostatic spray application (ESTA), for example, alone or in conjunction with hot spray application such as hot air spraying, for example. Application may take place at temperatures of max. 70 to 80°C, so that appropriate application viscosities are attained without any change or damage to the clearcoat material and its overspray (which may be intended for reprocessing) during the short period of thermal stress. For instance, hot spraying may be configured in such a way that the clearcoat material is heated only very briefly in the spray nozzle or shortly before the spray nozzle.

The spray booth used for application may, for example, be operated with a circulation system, which may be temperature-controllable, and which is operated with an appropriate absorption medium for the overspray, an

example of such medium being the clearcoat material itself.

Where the clearcoat material of the invention includes
5 constituents (C) crosslinkable with actinic radiation,
application is made under illumination with visible
light with a wavelength of above 550 nm, or in the
absence of light. By this means, material alteration or
damage to the clearcoat material and to the overspray
10 is avoided.

In the context of the process of the invention, the
application methods described above may be used to
produce all coats FL and BL and also, if desired,
15 further coats of the multicoat system ML of the
invention.

In accordance with the invention, the clearcoat film
may be cured thermally and/or with actinic radiation in
20 dependence on its material composition.

Curing may take place after a certain rest period. This
period may have a duration of from 30 s to 2 h,
preferably from 1 min to 1 h, and in particular from
25 1 min to 30 min. The rest period is used, for example,
for leveling and devolatilization of the coating films
or for the evaporation of volatile constituents such as
solvents, water or carbon dioxide if the clearcoat

material has been applied using supercritical carbon dioxide as solvent (H). The rest period may be shortened and/or assisted by the application of elevated temperatures up to 80°C, provided this does 5 not entail any damage or alteration to the coating films, such as premature complete crosslinking, for instance.

The thermal curing has no special features in terms of 10 its method but instead takes place in accordance with the customary and known methods such as heating in a forced-air oven or irradiation with IR lamps. Thermal curing may also take place in stages. Advantageously, it is effected at a temperature of from 50 to 100°C, 15 with particular preference from 80 to 100°C, and in particular from 90 to 100°C, for a period of from 1 min to 2 h, with particular preference from 2 min to 1 h, and in particular from 3 min to 30 min. Where the substrates used have a high capacity to withstand 20 thermal stress, thermal crosslinking may also be conducted at temperatures above 100°C. In general it is advisable in this case not to exceed temperatures of 160°C, preferably 140°C, and in particular 130°C.

25 Given an appropriate material composition of the clearcoat material, the thermal curing may be supplemented by curing with actinic radiation, it being possible to use UV radiation and/or electron beams. If

desired, it may be supplemented by or conducted with actinic radiation from other radiation sources. In the case of electron beams, it is preferred to operate under an inert gas atmosphere. This may be ensured, for 5 example, by supplying carbon dioxide and/or nitrogen directly to the surface of the coating film.

In the case of curing with UV radiation, as well, it is possible to operate under inert gas in order to prevent 10 the formation of ozone.

Curing with actinic radiation is carried out using the customary and known radiation sources and optical auxiliary measures. Examples of suitable radiation 15 sources are high or low pressure mercury vapor lamps, with or without lead doping in order to open up a radiation window of up to 405 nm, or electron beam sources. The arrangement of these sources is known in principle and may be adapted to the circumstances of 20 the workpiece and the process parameters. In the case of workpieces of complex shape such as automobile bodies, the regions not accessible to direct radiation (shadow regions) such as cavities, folds and other structural undercuts may be cured using point, small- 25 area or all-round emitters, in conjunction with an automatic movement means for the irradiation of cavities or edges.

The equipment and conditions for these curing methods are described, for example, in R. Holmes, U.V. and E.B. Curing Formulations for Printing Inks, Coatings and 5 Paints, SITA Technology, Academic Press, London, United Kingdom, 1984.

Curing may take place in stages, i.e., by multiple exposure to light or actinic radiation. This may also 10 be done alternately, i.e., by curing in alternation with UV radiation and electron beams.

Where thermal curing and curing with actinic radiation are employed together (dual cure), these methods may be 15 used simultaneously or in alternation. Where the two curing methods are used in alternation, it is possible, for example, to begin with thermal curing and end with actinic radiation curing. In other cases it may prove advantageous to begin and to end with actinic radiation 20 curing. The skilled worker is able to determine the curing method particularly appropriate to each individual case on the basis of his or her general knowledge in the art, possibly with the aid of simple preliminary experiments.

25

In the context of the process of the invention, these curing methods may be used to produce all coats FL and

BL and also, if desired, further coats of the multicoat system ML of the invention.

The resultant clearcoats KL of the invention have a 5 film thickness of from 10 to 100, preferably from 15 to 80, with particular preference from 20 to 70, and in particular from 25 to 60 μm .

The clearcoats KL of the invention are the essential 10 constituents of the multicoat system ML of the invention.

The multicoat systems ML of the invention may be produced in a variety of ways in accordance with the 15 process of the invention.

In a first preferred variant, the process of the invention comprises the following steps:

20 (I) preparing a basecoat film by applying a basecoat material to the substrate,

(II) drying the basecoat film,

25 (III) preparing a clearcoat film by applying a clearcoat material to the basecoat film, and

(IV) jointly curing the basecoat film and the clearcoat film to give the color and/or effect basecoat BL and the clearcoat KL (wet-on-wet technique).

5

A second preferred variant of the process of the invention comprises the steps of:

10 (I) preparing a surfacer film by applying a surfacer to the substrate,

15 (II) curing the surfacer film to give the surfacer coat FL,

20 (III) preparing a basecoat film by applying a basecoat material to the surfacer coat FL,

(IV) drying the basecoat film,

25 (V) preparing a clearcoat film by applying a clearcoat material to the basecoat film, and

(VI) jointly curing the basecoat film and the clearcoat film to give the color and/or effect basecoat BL and the clearcoat KL (wet-on-wet technique).

Which of the preferred variants is chosen depends on the intended use of the multicoat systems ML of the invention. For instance, the second variant, in particular, is employed with great preference in the 5 context of automotive OEM finishing.

Aqueous coating materials used to produce surfacer coats comprise as binders, for example, water-soluble or -dispersible polyesters and/or polyurethanes. 10 Aqueous coating materials of this kind are known from patent DE-A-43 37 961, DE-A-44 38 504, DE-C-41 42 816 or EP-A-0 427 028.

Suitable coating materials for producing the color 15 and/or effect basecoats BL are all customary and known basecoat materials, especially aqueous basecoat materials.

Examples of suitable basecoat materials, especially 20 aqueous basecoat materials, comprising acrylate copolymers as binders are known from patent DE-A-39 42 804, WO 99/15597 or DE-A-197 41 554.

Examples of suitable polyurethane-based aqueous 25 basecoat materials are known from patents EP-A-0 089 497, EP-A-0 256 540, EP-A-0 260 447, EP-A-0 297 576, WO 96/12747, EP-A-0 523 610, EP-A-0 228 003, EP-A-0 397 806, EP-A-0 574 417,

EP-A-0 531 510,	EP-A-0 581 211,	EP-A-0 708 788,
EP-A-0 593 454,	DE-A-43 28 092,	EP-A-0 299 148,
EP-A-0 394 737,	EP-A-0 590 484,	EP-A-0 234 362,
EP-A-0 234 361,	EP-A-0 543 817,	WO 95/14721,
5 EP-A-0 521 928,	EP-A-0 522 420,	EP-A-0 522 419,
EP-A-0 649 865,	EP-A-0 536 712,	EP-A-0 596 460,
EP-A-0 596 461,	EP-A-0 584 818,	EP-A-0 669 356,
EP-A-0 634 431,	EP-A-0 678 536,	EP-A-0 354 261,
EP-A-0 424 705,	WO 97/49745,	WO 97/49747,
10 EP-A-0 401 565, EP-B-0 730 613, and WO 95/14721.		

Accordingly, the multicoat systems ML of the invention may have a different structure.

15 In a first preferred variant of the multicoat system ML of the invention, the following coats are situated above one another in the stated sequence:

(1) a surfacer coat FL which absorbs mechanical
20 energy,

(2) a color and/or effect basecoat BL, and

(3) the clearcoat KL of the invention.

25

In a second preferred variant of the multicoat system ML of the invention, the following coats are situated above one another in the stated sequence:

(1) a color and/or effect basecoat BL, and

(2) the clearcoat KL of the invention.

5

In this case, the clearcoat KL may further be provided with a highly scratch-resistant coating such as, for example, an organically modified ceramic material.

10 In the process of the invention, the surfacer film and basecoat film are applied in a wet film thickness such that their curing gives coats FL and BL having the film thicknesses which are advantageous and necessary for their functions. In the case of the surfacer coat FL,
15 this thickness is from 10 to 150, preferably from 15 to 120, with particular preference from 20 to 100, and in particular from 25 to 90 μm , and in the case of the basecoat BL it is from 5 to 50, preferably from 10 to 40, with particular preference from 12 to 30, and in
20 particular from 15 to 25 μm .

The multicoat systems ML of the invention, owing to the particularly advantageous properties of the clearcoats KL of the invention, have an outstanding profile of
25 properties which is very well balanced in terms of mechanical properties, optical properties, corrosion resistance, hardness, and adhesion. Thus the multicoat

systems ML of the invention have the intercoat adhesion and high optical quality which the market requires and no longer give rise to any problems such as inadequate condensation resistance, cracking (mud cracking) in the 5 basecoats BL, or leveling defects or surface structures in the clearcoats KL. In particular, the multicoat system ML of the invention has an outstanding D.O.I. (depth of image) and outstanding surface smoothness.

10 Not least, however, it proves to be a very particular advantage that by means of the clearcoat material of the invention and of the process of the invention it is possible in a simple manner to produce a multicoat system ML which is based exclusively on aqueous coating 15 materials and, if appropriate, on pulverulent basecoat materials.

Examples

Preparation Example 1

5 The preparation of a dispersion of a copolymer (A)

A steel reactor as commonly used to prepare dispersions, equipped with a stirrer, a reflux condenser and 3 feed vessels, was charged with 52.563 parts by weight of DI water and this initial charge was heated to 90°C. The first feed vessel was charged with 10.182 parts by weight of acrylic acid, 18.345 parts by weight of methyl methacrylate and 1.493 parts by weight of diphenylethylene. The second feed vessel was charged with 9.914 parts by weight of 25 percent strength ammonia solution. The third feed vessel was charged with 5.25 parts by weight of DI water and 2.253 parts by weight of ammonium peroxodisulfate. With intensive stirring of the initial charge in the steel reactor, the three feed streams were commenced simultaneously. The first and second feed streams were metered in over the course of one hour. The third feed stream was metered in over the course of 1.25 hours. The resultant reaction mixture was held at 90°C for 4 hours and then cooled to below 40°C and filtered through a 100 µm GAF bag. The resultant dispersion had a solids content of from 32 to 34% by weight (1 hour, 130°C) and a free

monomer content of less than 0.2% by weight (determined by gas chromatography).

The dispersion (A) was used to prepare a block
5 copolymer (A).

Preparation Example 2

The preparation of a dispersion of a block copolymer

(A)

5

A steel reactor as commonly used to prepare dispersions, equipped with a stirrer, a reflux condenser and a feed vessel, was charged with 51.617 parts by weight of DI water and 9.907 parts by weight 10 of the dispersion from Example 1 and this initial charge was heated to 90°C with stirring. Thereafter, a mixture of 9.856 parts by weight of n-butyl methacrylate, 7.884 parts by weight of styrene, 12.661 parts by weight of hydroxyethyl methacrylate and 8.885 15 parts by weight of ethylhexyl methacrylate was metered in from the feed vessel over the course of six hours. The resultant reaction mixture was stirred at 90°C for two hours. Subsequently, the resultant dispersion was cooled to below 40°C and filtered through a 50 µm GAF 20 bag. The dispersion (A) had a solids content of from 41 to 42% by weight (1 hour, 130°C) and a free monomer content of less than 0.2% by weight (determined by gas chromatography).

25 **Example 1**

Th preparation of the clearcoat material of the
invention

The clearcoat material of the invention was prepared
5 from 100 parts by weight of the dispersion (A) from
Preparation Example 2, 5 parts by weight of a standard
commercial crosslinking agent based on
tris(alkoxycarbonylamino)triazines (Cylink® 2000;
CYTEC) and 0.4 part by weight of Agitan® 281. The low-
10 viscosity mixture was homogenized with an Ultraturrax.
Thereafter, the viscosity was 128 mPa.s at a shear rate
of 1.000 s⁻¹ and at 23°C.

Example 2

The production of a clearcoat KL of the invention

5 The clearcoat KL of the invention was produced using customary and known steel test panels which have been coated with an electrodeposition coat produced from a standard commercial electrodeposition coating material and with a surfacer coat produced from a standard
10 commercial surfacer.

The clearcoat material of the invention from Example 1 was applied to the surfacer coat of the test panels with the aid of a pneumatic spray gun. The resultant
15 clearcoat film was baked at 140°C for 30 minutes.

This gave smooth transparent clearcoats KL with a film thickness of 40 µm, which were free from surface defects and gel specks. They were fully crosslinked and
20 could not be damaged or even removed by more than 100 MEK double strokes. The clearcoats KL withstood 240 hours of artificial weathering under UV light in accordance with the sun test (conducted with the apparatus "SUN-Test" from Heraeus with an NXE 1500 B
25 emitter) without being damaged.